



Effects of cathode materials on H₂O₂ production in microbial fuel cells

Pin-Hsueh Wu^a, Yinhan Wang^b, Pei-Hsun Wu^a, Shih-Yuan Lu^c, Chang-Ping Yu^{a,*}

^aGraduate Institute of Environmental Engineering, National Taiwan University, Taipei 10617, Taiwan, Tel. +886-2-3366-3729, email: cpyu@ntu.edu.tw (C.-P. Yu)

^bXiamen Institute of Rare-earth Materials, Haixi Institutes, Chinese Academy of Sciences, Xiamen, Fujian 361021, China

^cDepartment of Chemical Engineering, National Tsing Hua University, Hsinchu30013, Taiwan

Received 5 June 2018; Accepted 3 March 2019

ABSTRACT

The application of microbial fuel cells (MFCs) on wastewater treatment has been evolving rapidly in recent years. It provides simultaneous pollutants removal as well as energy recovery in treating wastewater. MFCs can also be needed to produce H₂O₂ and drive electro-Fenton reaction to remove emerging contaminants in wastewater. However, the cell voltage produced in a MFC is lower than the voltage used to drive the conventional electro-Fenton system. Therefore, this study intended to explore the H₂O₂ formation reactions on the cathode materials of the conventional electro-Fenton system in the MFC. In this study, H₂O₂ production kinetics on different cathode materials, including the non-catalytic materials such as graphite and steel plate, the high specific surface area materials such as carbon nanotubes, activated carbon, the catalytic materials such as platinum-titanium plate and the high specific surface area catalytic materials such as manganese oxide/carbon aerogel were evaluated. These cathode materials showed different H₂O₂ production performance. Among them, the manganese oxide/carbon aerogel with high specific surface area and catalytic activity demonstrated the best performance of H₂O₂ production among testing materials. The reaction kinetics of H₂O₂ production rates on high specific surface area cathode materials, the activated carbon and carbon nanotubes, were found to be pseudo-first order. However, for the cathode materials with catalytic surface, the manganese oxide/carbon aerogel and platinum titanium, the reaction kinetics of H₂O₂ production rates were found to be pseudo-zero order within 12 h. Our results suggest that the catalytic cathode materials with high specific surface area, like manganese oxide/carbon aerogel, may be more promising for H₂O₂ production in MFCs to provide the electro-Fenton system. Therefore, the development of better cathode materials should be focused for the future application of MFCs driven electro-Fenton system.

Keywords: Microbial fuel cells (MFCs); Hydrogen peroxide; High specific surface area materials; Catalytic materials; Kinetics

*Corresponding author.